

ACID-BASE BALANCE IN A MARINE TELEOST: (Myoxocephalus Octodecimspinosus)  
THE EFFECT OF  $\text{NH}_4\text{Cl}$  INFUSION

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Teleost fish excrete nitrogenous waste products in the form of ammonia ( $\text{NH}_3 + \text{NH}_4^+$ ). The majority of ammonia transfer to the surrounding water is via the gills. It is not known whether ammonia is excreted by non-ionic diffusion of  $\text{NH}_3$ , the electroneutral exchange of serosal  $\text{NH}_4^+$  for external  $\text{Na}^+$ , an ionic diffusive efflux of  $\text{NH}_4^+$ , or some combination of these mechanisms. In the freshwater adapted trout (Salmo gairdneri) diffusive loss of  $\text{NH}_3$  has been shown to be the principle route of ammonia excretion under normal conditions (Cameron & Heisler, J. Exp. Biol. 105:107-125, 1983).  $\text{Na}^+/\text{NH}_4^+$  exchange and  $\text{NH}_4^+$  diffusion are thought to play a role in ammonia efflux in seawater species (Claiborne et al., J. Exp. Biol. 96:431-434, 1982; Goldstein et al., J. Exp. Zool. 219:395-398, 1982). As demonstrated by Cameron and Kormanik (J. Exp. Biol. 99:143-160, 1982) and Cameron and Heisler (1983, Ibid.), the infusion of an ammonium salt ( $\text{NH}_4\text{Cl}$  or  $(\text{NH}_4)_2\text{SO}_4$ ) into either the freshwater catfish (Ictalurus punctatus) or the S. gairdneri induces a pronounced plasma acidosis. The authors ascribe this effect to the rapid loss of the infused ammonia as  $\text{NH}_3$ , with a more gradual equimolar excretion or intracellular buffering of the remaining  $\text{H}^+$ , thus inducing the rapid decrease and secondary recovery of the observed plasma pH. To gain additional insight into the possible role of  $\text{NH}_3$  diffusion in a marine teleost, it was of interest to duplicate these experiments by subjecting a stenohaline marine species to an infusion of  $\text{NH}_4\text{Cl}$  while monitoring extracellular acid-base status and ammonia excretion.

Long-horned sculpin (Myoxocephalus octodecimspinosus, mass =  $182 \pm 6$  grams ( $n = 9$ ); mean  $\pm$  S.E.) were anesthetized (ms-222, 1:10,000) and cannulated in a manner similar to that described for the measurement of ventral aortic blood pressure (Claiborne & Evans, Mar. Biol. Lett. 2:123-130, 1981). In the present study, the cut tip of a 26 gauge needle connected to short length of heparinized, Ringer's-filled cannula (PE-50) was inserted into the afferent artery of the third branchial arch and secured in place with two sutures. An additional Ringer's-filled cannula (PE-50) was inserted through the skin and peritoneal musculature into the peritoneal cavity. The sculpin were then placed in a darkened plexiglass box (vol = 2 l) and allowed to recover for 24-48 hours. During this period, fresh running seawater ( $12-15^\circ\text{C}$ ) was directed through the experimental chamber. Just prior to the start of the experimental period, duplicate control blood samples (0.5 mls) were collected via the aortic cannula. The experiment was begun when  $\text{NH}_4\text{Cl}$  (5 mM/kg, a 4-5 ml bolus of 200 mM  $\text{NH}_4\text{Cl}$ ) was infused intraperitoneally into the animal over several minutes. After a 5 minute equilibration, a time zero blood sample was drawn, and additional samples were obtained at hours 1, 2, 4, 8, and 20 post-infusion. Blood pH and  $\text{TCO}_2$  were measured utilizing a blood-gas analyzer (I.L. model 213) and a  $\text{TCO}_2$  detection system (Capni-Con II; Cameron Instruments Inc.). Plasma  $\text{HCO}_3^-$  and  $\text{pCO}_2$  were calculated from pH and  $\text{TCO}_2$  using values for  $\alpha\text{CO}_2$  and  $\text{pK}'$  at  $15^\circ\text{C}$  calculated from those for human plasma reported by Severinghaus (In "Handbook of Physiology", Section 3, Respiration, American Physiological Society, 1965). Plasma total ammonia ( $\text{NH}_3 + \text{NH}_4^+$ ) was measured enzymatically (Sigma kit #170-UV). Three-four hours prior to the beginning of an experiment, the running seawater was disconnected so that changes in the bath concentration of total

ammonia [ $T_{amm}$ ] could be monitored. Water samples (15 mls) were collected prior to, and during the experiment, and analyzed for [ $T_{amm}$ ] using the phenolhypochlorite method. Four or eight hours post-infusion, the water in the fish chamber was briefly flushed with fresh seawater to limit the accumulation of external ammonia. [ $T_{amm}$ ] excretion (mM/kg·hr) was calculated from the changes in water [ $T_{amm}$ ] after adjusting for volume changes, the mass of the animal, and time between samples.

The effect of  $NH_4Cl$  infusion on internal acid-base status in the sculpin is shown in Table 1. As would be expected,  $NH_4Cl$  infused into the peritoneal cavity induced a large increase in the [ $T_{amm}$ ] of the blood which persisted for 1-2 hours. A rapid plasma pH depression of more than 0.4 units was observed immediately following the  $NH_4Cl$  infusion. Plasma pH then returned to control values within 4 hours post-infusion. The drop in pH was due to a decrease in serosal [ $HCO_3^-$ ] (or an increase in [ $H^+$ ]) and an elevation in  $PCO_2$ . After 20 hours, plasma [ $HCO_3^-$ ] had increased by  $\approx 50\%$  over control measurements, thus compensating for the additional blood  $CO_2$ . Net  $T_{amm}$  efflux from the fish to the water was increased by  $\approx 8x$  during the first 2 hours post-infusion.

It is clear that  $NH_4Cl$  infusion elicits pronounced internal acid-base effects. Indeed, the alterations in extracellular pH, [ $HCO_3^-$ ], and  $PCO_2$  are similar to those found following an acid infusion (Cameron & Kormanik, 1982, Op. Cit.). During the first two hours after the infusion, the  $\Delta T_{amm}$  efflux (the net experimental rate - the net control rate) was 3.33 mM/kg. In other words, 2/3 of the infused ammonia was lost to the water concurrent to the observed plasma pH depression. This is a good indication that a significant portion of the  $T_{amm}$  infused as  $NH_4Cl$  was transferred to the water as  $NH_3$ . The remaining protons thus caused the observed acid-base effects. It is likely that over the next few hours the additional  $H^+$  was excreted to the water (via  $Na^+/H^+$  exchange, see Evans, J. Exp. Biol. 97:289-299, 1982), compensated by an uptake of  $HCO_3^-$  from the water by  $Cl^-/HCO_3^-$  exchange (Claiborne & Heisler, J. Exp. Biol. 108:25-43, 1984), or buffered intracellularly. Goldstein et al. (1982, Op. Cit.) found that a 10x increase in the Ringer's [ $NH_3$ ] did not stimulate the  $T_{amm}$  efflux from the isolated, perfused head of the sculpin, thus indicating a low transbranchial  $NH_3$  permeability in this preparation. The reason for the discordance between these *in vitro* results and our data is unknown at present. It also remains to be seen whether sculpin excrete  $NH_3$  during "control" conditions, but it is clear that at least one species of marine teleost is capable of  $NH_3$  excretion when faced with the stress imposed by high internal  $T_{amm}$ . (Funded by NSF PCM 83-02621 to DHE and an intramural grant from Georgia Southern College to JBC.)

Table 1. The effect of ammonium salt infusion on extracellular acid-base status and ammonia efflux in *M. octodecimspinosus*.

Period	[ammonia] (mM/l)	pH	[ $HCO_3^-$ ] (mM/l)	$pCO_2$ (torr)	$J_{amm}$ (mM/kg·hr)
control:	$1.46 \pm 0.54$ (N = 6)	$7.835 \pm 0.048$ (n = 7)	$4.80 \pm 0.54$ (n = 7)	$1.93 \pm 0.33$ (n = 7)	$0.24 \pm 0.03$ (n = 8)
post-infusion: (hours)					
0	$4.91 \pm 0.94$ *	$7.423 \pm 0.102$ **	$2.98 \pm 0.55$ **	$3.08 \pm 0.38$ *	$2.68 \pm 0.18$ *
1	$1.77 \pm 0.34$	$7.515 \pm 0.072$ **	$3.29 \pm 0.42$ **	$3.13 \pm 0.60$ *	$1.13 \pm 0.11$ *
2	$1.09 \pm 0.07$	$7.682 \pm 0.040$ **	$5.74 \pm 0.63$	$2.98 \pm 0.35$ *	$0.32 \pm 0.06$
4	$0.51 \pm 0.10$	$7.809 \pm 0.028$	$6.54 \pm 0.88$ *	$2.63 \pm 0.29$	$0.48 \pm 0.06$
8	$0.82 \pm 0.25$	$7.767 \pm 0.030$	$5.88 \pm 0.69$ *	$2.64 \pm 0.25$	$0.48 \pm 0.06$
20	$1.36 \pm 0.42$	$7.787 \pm 0.025$	$7.34 \pm 0.92$ *	$3.21 \pm 0.38$ *	$0.20 \pm 0.04$

Mean  $\pm$  S.E. test of significance by paired t-test,  $p < 0.05$ , two tailed: \* = significant increase from control values; \*\* = significant decrease from control values.  $J_{amm}$  = net ammonia efflux calculated for each sampling interval.