

from that in mammalian blood. Plasma  $\text{Na}^+$  and  $\text{Cl}^-$  concentrations are significantly higher while  $\text{K}^+$ ,  $\text{H}^+$ ,  $\text{H}_2\text{CO}_3$  and  $\text{HCO}^-$  are significantly lower. No data are available concerning the concentrations of these electrolytes in red cell intracellular water. Simultaneous determinations of plasma and red cell concentrations of  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Cl}^-$  and  $\text{H}^+$  (DMO technique) were performed in 10 dogfish. The ratio of intracellular to plasma concentrations were calculated. Mean values are as follows:

	(RBC) meq/Kg $\text{H}_2\text{O}$	Plasma meq/Kg $\text{H}_2\text{O}$	$\frac{[\text{C}]}{[\text{C}]} \frac{\text{RBC}}{\text{Plasma}}$
$\text{Cl}^-$	141	248	0.566
$\text{H}^+$	$6.0 \times 10^{-8} \text{ M/L}$	$3.2 \times 10^{-8} \text{ M/L}$	0.530 *
$\text{Na}^+$	43	246	0.183
$\text{K}^+$	211	3.3	72.2

$$* \frac{[\text{H}^+]}{[\text{H}^+]} \frac{\text{Plasma}}{\text{RBC}}$$

These data show that  $\text{Cl}^-$  and  $\text{H}^+$ , as is probably true of mammalian red cells, are passively distributed obeying Gibbs-Donnan distribution since  $\frac{[\text{H}^+]}{[\text{H}^+]} \frac{\text{RBC}}{\text{Plasma}} = \frac{[\text{Cl}^-]}{[\text{Cl}^-]} \frac{\text{Plasma}}{\text{RBC}}$ . The dogfish red cell resembles that of man and differs from the dog red cell in being a low  $\text{Na}^+$ , and high  $\text{K}^+$  cell.

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#### INTRACELLULAR ELECTROLYTE PATTERNS IN THE DOGFISH. II. MUSCLE AND BRAIN

H. V. Murdaugh, E. D. Robin, P. Soteres, and E. Weiss, University of Alabama, Birmingham, Ala., and University of Pittsburgh, Pittsburgh, Pa.

The extracellular electrolyte pattern of elasmobranchs differs in striking fashion from that of mammals. No data are available concerning the electrolyte pattern of intracellular water. Therefore measurements of  $\text{Na}^+$ ,  $\text{K}^+$  and  $\text{Cl}^-$  concentrations in muscle and brain water of the dogfish, *S. acanthias*, were performed and have been compared with corresponding values in the extracellular fluid.

Measurements were performed on tissues obtained from 10 animals and the following results were obtained:

	$\text{Na}^+$ meq/Kg $\text{H}_2\text{O}$	$\text{K}^+$ meq/Kg $\text{H}_2\text{O}$	$\text{Cl}^-$ meq/Kg $\text{H}_2\text{O}$
Plasma	246	3.33	248
"ECF"	234	3.16	260
Muscle	57	162.00	35
Brain	136	146.00	98

In the case of muscle it would appear that the volume of ECF is quite low, the chloride space averaging only 13% (concentration per volume of  $H_2O$  and not weight). Elasmobranch muscle cells are low in  $Na^+$  and high in  $K^+$  suggesting that the general mechanism of maintaining ion gradients in muscle is the same in the elasmobranch as in mammals.

In the case of brain, the mean chloride space is approximately 3 times that of muscle and it would appear that some of brain chloride is intracellular in location. In addition the distribution of values of  $Na^+$  and  $K^+$  appear to show 2 populations of cells some of which are rich in  $Na^+$  and  $Cl^-$  and some of which are rich in  $K^+$ . This dual population has also been found in mammalian brain cells and it has been suggested that the  $NaCl$ -rich cells are glial; while the  $NaCl$ -poor cells are neuronal.

In general, these data suggest that although the absolute concentrations of these ions differ markedly from the mammalian pattern, ratios between ECF and ICF are similar suggesting basically functionally similar mechanisms for maintaining ion gradients.

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#### PROBABLE FUNCTION OF THE ALKALINE COELOMIC FLUID OF THE FRESHWATER TURTLE, *Pseudemys scripta elegans*

H. V. Murdaugh, E. D. Robin, W. Pyron, and E. Weiss, University of Alabama, Birmingham, Ala., and University of Pittsburgh, Pittsburgh, Pa.

In 1929 Homer Smith demonstrated that coelomic fluid in the freshwater turtle is more alkaline than plasma and contains a higher concentration of  $HCO_3^-$  (circa 100 mM/L as compared with 35 mM/L in plasma). No definitive function has been described for this fluid.

Recent studies in this laboratory have demonstrated that during diving the turtle uses anaerobic metabolism for energy and as a result generates large amounts of lactic acid. The ability of lactic acid to penetrate coelomic fluid was compared with the penetration by mineral acid (HCl). Coelomic fluid pH and  $HCO_3^-$  concentrations were measured for periods up to 5 hours following the administration of approximately 5 meq of  $H^+$  in the form of either lactic or hydrochloric acid. These studies showed that mineral acid penetrates coelomic fluid quite slowly as compared with the rapid penetration of lactic acid. Preliminary measurements of lactic acid concentrations in plasma and coelomic fluid suggest that either equilibrium was not attained in the period of study or that simple non-ionic diffusion does not explain the penetration of lactic acid into coelomic fluid. However, it is clear that a substantial amount of infused lactic acid is buffered in this compartment. It is possible that a function of the alkaline coelomic fluid is to buffer endogenously generated lactic acid arising from anaerobiosis during prolonged diving.

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